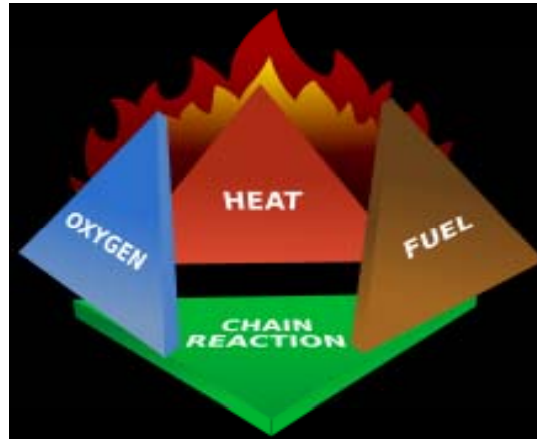


# PHYSICS OF FIRE

Fire energy and release rates

Mark Fishlock

# FIRE (Combustion)



FIRE : a state, process, or instance of combustion in which FUEL or other material is ignited and combined with OXYGEN, giving off light, heat, flame. and various reaction products.

COMBUSTION: any process in which a substance reacts with oxygen to produce a significant rise in temperature and the emission of light

# HEAT (energy)

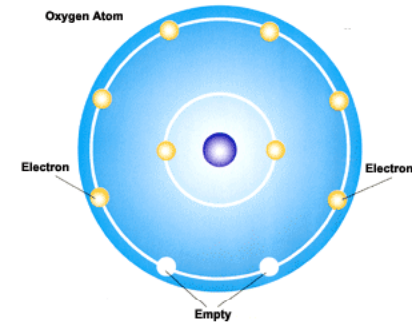
True Combustion is an Exothermic reaction (gives off heat) and is thus self perpetual



The energy given off from a fire is measured in  
Watts



# OXYGEN - O<sub>2</sub>



- Abundant in the atmosphere at @ 21%
- Below 18.5% fire activity is greatly suppressed
- Below 16% it ceases.
- Oxygen rich environment is defined as an environment in which the concentration of oxygen is greater than 25% (at STP)
- Combustion has no upper O<sub>2</sub> limit (16-100%)

# OXYGEN - O<sub>2</sub>

Pressure and the atmosphere

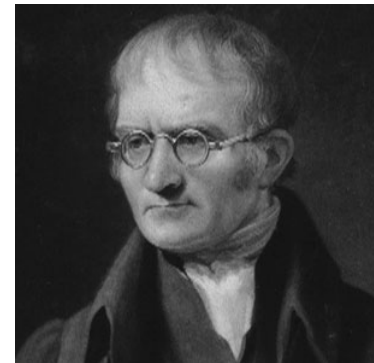
Atmospheric pressure = 1.01325 Bar

Partial pressure relates to each component of a gas

For Air this approximates to:

$$n = n_{\text{nitrogen}} + n_{\text{oxygen}}$$

$$= 0.213 \text{ Bar Oxygen pressure (3psi)}$$



# Oxygen rich Fires

Tend to burn at @ $2x^2$  rate  
(Double the O<sub>2</sub> = 16x faster)

Ignition temperatures change

Flammable limits change

Hard to extinguish



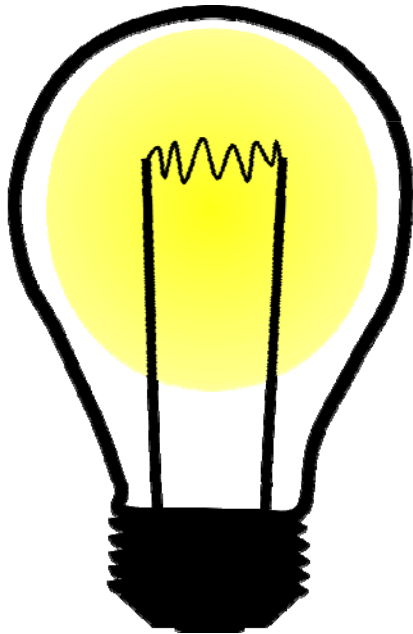
# FUEL

- Rated by its potential to release its energy
- It has a Calorific value measured in Joules (complete combustion)
- The energy Release Rate is speed at which fuel is burnt
- Release rate is normally expressed in normal free air (20.946% Oxygen)



# FUEL :Release rate

A Joule is the equivalent of 1 watt for 1 second.



100w Light bulb burning for 10 hours uses:

$$100 \times 3600 = 360000\text{J}$$

Or 360KJ (Kilojoules)

Or 0.36MJ (Mega joules)

Or 0.00036GJ (Gigajoules)

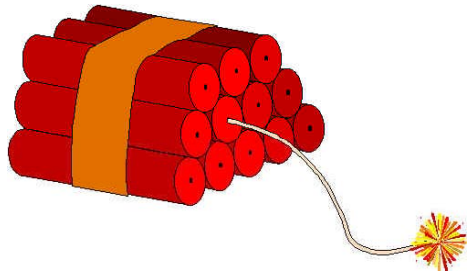


# FUEL: Release rate

A Tonne of TNT is a unit of energy.

Calorific value equal to

4.184 gigajoules,



A Tonne of Coal (anthracite)

Calorific value equal to

32.5 gigajoules



# FUEL: Release rate

A Tonne of TNT in a square cube 0.6m x 0.6m will burn in 0.00001 of a second



A Tonne of Coal (anthracite) in a cube 1.6m x 1.6m will burn in 8 hours

# FUEL: Release rate

Most affected by the fuels ability to use Oxygen:

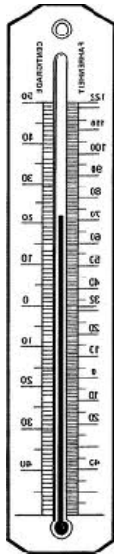
Surface area

Pressure

Temperature

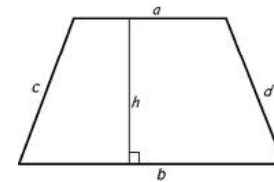
Mix ratios

Flame disruption



Trapezoid

Perimeter  $p = a + b + c + d$



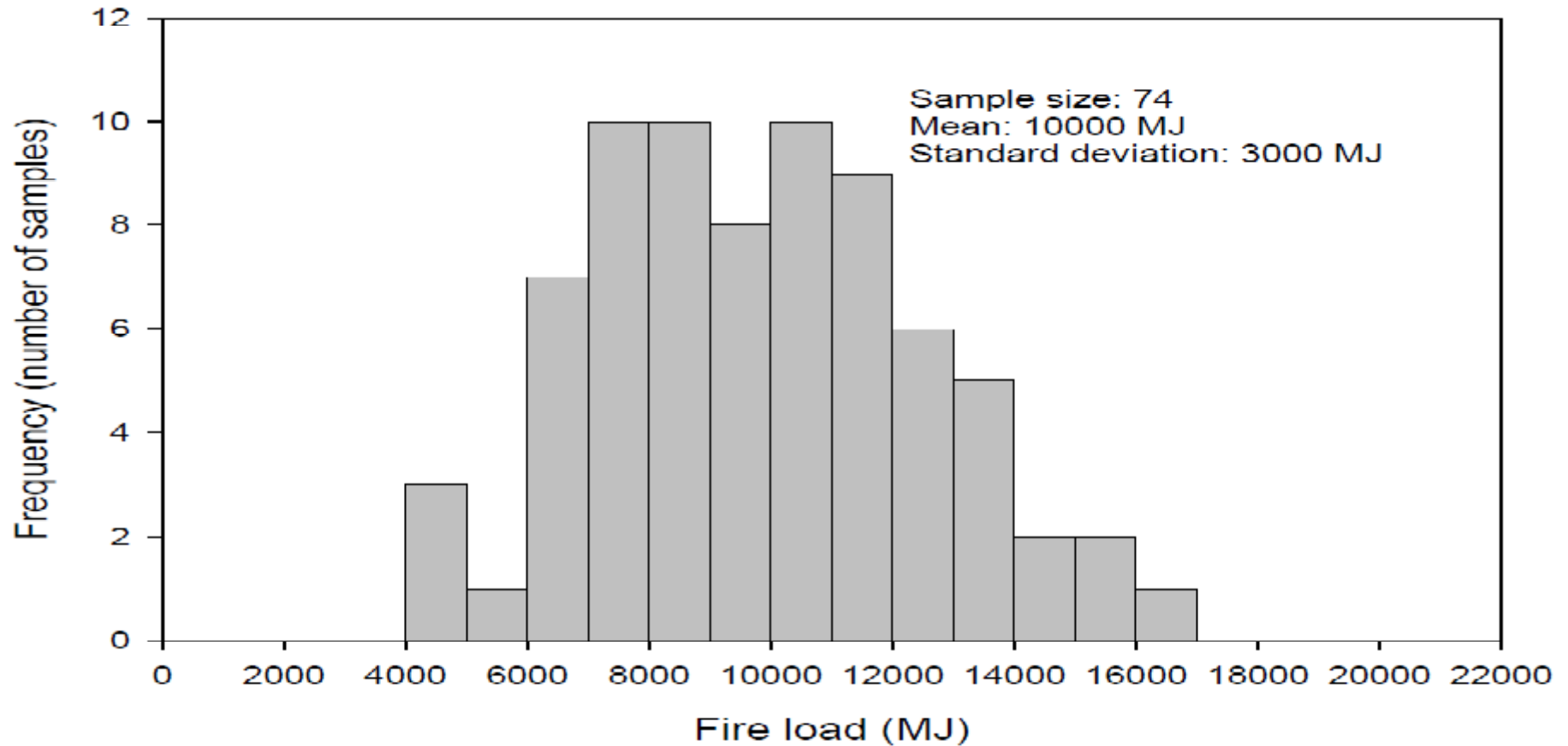
Area  $A = \frac{(a+b)h}{2}$  or  $A = \frac{1}{2}(a+b)h$



# Peak Heat Release Rates

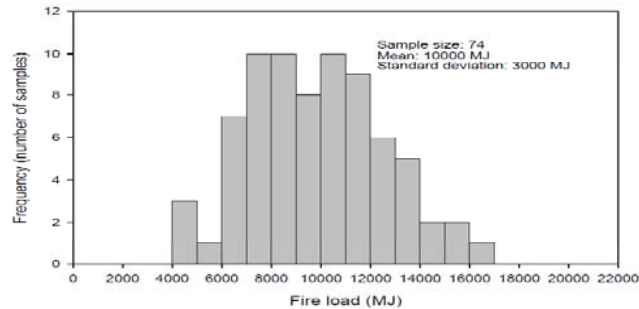
- **CFBT Container Simulation - 1.5MW**
  - **Sofa (2-seater) - 3.0MW**
  - **Sofa (3-seater) - 3.5MW**
  - Upholstered Chair - 2.0MW
  - Rubbish Bin (small) - 300KW
  - Light Bulb - 100W
  - Xmas Tree - 0.7MW
  - Small Dresser - 1.8MW
  - Single Mattress - 1.0MW
  - **Pine Bunk Beds - 4.5MW**
  - 5 Timber Pallets - 1.8MW
  - 2 Panel Work Station - 1.8MW
  - **TV & Video with 40 VHS tapes – 4MW**
  - Kings Cross Fire - 15-25MW
- 
- NIST wind driven flat fire simulation (25mph) 16.7 MW

# FUEL: Loading



2007 study by NRCC Building research

# FUEL: Loading



Potential fuel load 10,000 MJ

So if this buildings contents burnt un-aided for 60 mins it would liberate a constant average of 2.7 MW of energy